Date Name Group Group
Lab report form for the practical lesson on biochemistry
Topic: Basic organic reactions. Properties of proteins. Electrophoresis
Task 1 - Electrophoresis of serum proteins in 0.5% agarose
Principle:
Result:
Describe or make a drawing of the protein bands that you can see on the stained agarose gel.

Task 2: Basic organic reaction

2.1 Oxidation of alcohols

Principle:

Write <u>structural formulas</u> of all three alcohols and the scheme of their oxidation.

Results:

	A	В	С	W
Final color after reaction				
with KMnO ₄				

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	'n	n	$\boldsymbol{\alpha}$	11	CI	on	•

2.2 Coupling reactions of diazonium salts

Principle:Use <u>structural formulas</u>

Results:

Conclusion:

2.3 Reactions of carbonyl group

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Provide a detailed description of chemistry of each of the used test.

Results:

Write down your results (pos./neg., and record also the resulting color) of all tests for individual samples in the table provided. Conclude whether the obtained results meet theoretical expectations.

	Acetone	Formaldehyde	Acetic acid	Formic acid	Water
Legal		X		X	X
Fehling	X			X	X
Tollens	X		X		
Schiff	X			X	

Conclusion:

2.4 Esterification

Principle:
Using <u>structural formulas</u> write down the equation of esterification you performed.
Observation:
Observation:
Conclusion:
Task 3: Reversible precipitation of proteins
3.1 Precipitation of egg-white protein with sodium chloride and its reversal
3.1 Precipitation of egg-white protein with sodium chloride and its reversal Principle:
Principle:
Principle:
Principle: Results/Observations:

Task 4: Precipitation of proteins by denaturation

Principle: What is denaturation?
Result/observation: 4.1 Precipitation of proteins with heavy metal salts
4.2 Precipitation of proteins with mineral acids
4.3 Precipitation of proteins with organic acids
4.4 Precipitation of proteins with high temperature (boiling)
Conclusion: Summarize what agents caused protein denaturation. Why in certain conditions denaturation did not lead to protein precipitation?