

Date Name Group

Lab report for the practical lesson on biochemistry

Topic: Examination of liver and pancreas

Task 1: Estimation of aspartate aminotransferase (AST) in serum

Principle:

(Use structural formulas and include also the indication reaction)

Results and calculations:

Time:		ΔA_{340}
0	A ₀	
1 minute	A ₁	A ₀ - A ₁ → ΔA_1
2 minutes	A ₂	A ₁ - A ₂ → ΔA_2
3 minutes	A ₃	A ₂ - A ₃ → ΔA_3

$$\Delta A_{340} = \frac{\Delta A_1 + \Delta A_2 + \Delta A_3}{3} = \dots\dots\dots$$

$$S\text{-AST } (\mu\text{kat/l}) = \Delta A_{340}/\text{min.} \times 36.2 = \dots\dots\dots$$

Conclusion:

Compare the catalytic concentration of AST in your sample with reference values.

Task 2: Estimation of alanine aminotransferase (ALT) in serum

Principle:

(Use structural formulas and include also the indication reaction)

Results and calculations:

Time:		ΔA_{340}
0	A_0	
1 minute	A_1	$A_0 - A_1 \rightarrow \Delta A_1$
2 minutes	A_2	$A_1 - A_2 \rightarrow \Delta A_2$
3 minutes	A_3	$A_2 - A_3 \rightarrow \Delta A_3$
4 minutes	A_4	$A_3 - A_4 \rightarrow \Delta A_4$
5 minutes	A_5	$A_4 - A_5 \rightarrow \Delta A_5$

$$\Delta A_{340} = \frac{\Delta A_1 + \Delta A_2 + \Delta A_3 + \Delta A_4 + \Delta A_5}{5} = \dots\dots\dots$$

$$S\text{-ALT } (\mu\text{kat/l}) = \Delta A_{340}/\text{min.} \times 29.5 = \dots\dots\dots$$

Conclusion:

Compare the catalytic concentration of ALT in your sample with reference values.

Task 3: Estimation of γ -glutamyl transferase (GGT) in serum

Principle:

(Use structural formulas)

Results and calculations:

Time:		ΔA_{405}
0	A_0	
1 minute	A_1	$A_1 - A_0 \rightarrow \Delta A_1$
2 minutes	A_2	$A_2 - A_1 \rightarrow \Delta A_2$
3 minutes	A_3	$A_3 - A_2 \rightarrow \Delta A_3$

$$\Delta A_{405}/\text{min.} = \frac{\Delta A_1 + \Delta A_2 + \Delta A_3}{3} = \dots\dots\dots$$

$$\text{GGT } (\mu\text{kat/l}) = \Delta A_{405}/\text{min.} \times 18.52 = \dots\dots\dots$$

Conclusion:

Compare the catalytic concentration of GGT in your sample with reference values.

Task 4: Estimation of serum amylase (AMS)

Principle:

Results and calculations:

Time:		ΔA_{405}
0	A_0	
1 minute	A_1	$A_1 - A_0 \rightarrow \Delta A_1$
2 minutes	A_2	$A_2 - A_1 \rightarrow \Delta A_2$
3 minutes	A_3	$A_3 - A_2 \rightarrow \Delta A_3$

$$\Delta A_{405} = \frac{\Delta A_1 + \Delta A_2 + \Delta A_3}{3} = \dots\dots\dots$$

$$S\text{-AMS } (\mu\text{kat/l}) = \Delta A_{405}/\text{min.} \times 54.9 = \dots\dots\dots$$

Conclusion:

Compare the catalytic concentration of AMS in your sample with reference values.